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## Hexaaquazinc(II) bis[4-(2-hydroxybenzylideneamino)benzenesulfonate]

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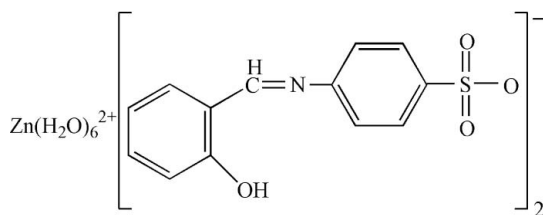
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Key indicators: single-crystal X-ray study;  $T = 298$  K; mean  $\sigma(\text{C}-\text{C}) = 0.010$  Å;  $R$  factor = 0.068;  $wR$  factor = 0.168; data-to-parameter ratio = 13.2.

In the title compound,  $[\text{Zn}(\text{H}_2\text{O})_6](\text{C}_{13}\text{H}_{10}\text{NO}_4\text{S})_2$ , a distorted  $\text{ZnO}_6$  octahedron results from the coordination by the six water molecules. Only three of the water molecules are crystallographically unique, as the Zn atom lies on an inversion center. The Zn—O bond lengths are in the range 2.054 (4)–2.073 (4) Å. A network of hydrogen bonds helps to establish the crystal packing.

## Related literature

For related literature, see: Tai *et al.* (2005).

## Experimental

## Crystal data

 $[\text{Zn}(\text{H}_2\text{O})_6](\text{C}_{13}\text{H}_{10}\text{NO}_4\text{S})_2$  $M_r = 726.03$ Monoclinic,  $P2_1/n$  $a = 6.3255$  (10) Å $b = 35.312$  (3) Å $c = 6.9832$  (10) Å $\beta = 90.391$  (2)° $V = 1559.8$  (4) Å<sup>3</sup> $Z = 2$ Mo  $K\alpha$  radiation $\mu = 0.99$  mm<sup>-1</sup> $T = 298$  (2) K $0.35 \times 0.33 \times 0.20$  mm

## Data collection

Bruker SMART CCD area-detector diffractometer

Absorption correction: multi-scan

(SADABS; Bruker, 1997)

 $T_{\min} = 0.723$ ,  $T_{\max} = 0.826$ 

6994 measured reflections

2708 independent reflections

2170 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.034$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.069$  $wR(F^2) = 0.168$  $S = 1.08$ 

2708 reflections

205 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.37$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.79$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O7—H7B $\cdots$ O3 <sup>i</sup>	0.85	1.93	2.760 (6)	166
O7—H7A $\cdots$ O1	0.85	1.93	2.779 (5)	177
O6—H6B $\cdots$ O1 <sup>ii</sup>	0.85	1.92	2.773 (6)	176
O6—H6A $\cdots$ O2 <sup>iii</sup>	0.85	1.92	2.770 (6)	175
O5—H5B $\cdots$ O3	0.85	1.90	2.745 (6)	171
O5—H5A $\cdots$ O2 <sup>ii</sup>	0.85	1.91	2.742 (6)	167
O4—H4 $\cdots$ N1	0.82	1.93	2.602 (7)	139

Symmetry codes: (i)  $x + 1, y, z$ ; (ii)  $x, y, z - 1$ ; (iii)  $x + 1, y, z - 1$ .

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PR2013).

## References

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**supplementary materials**

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## Hexaaquazinc(II) bis[4-(2-hydroxybenzylideneamino)benzenesulfonate]

X.-S. Tai, J. Yin and M.-Y. Hao

### Comment

As part of our ongoing studies of metal coordination complexes with Schiff base ligands (Tai *et al.*, 2005), the synthesis and structure of the title compound, (I), is reported. Six water molecules are attached to the zinc atom, resulting in a distorted  $ZnO_6$  octahedron (Fig. 1). The C7=N1 bond length [1.280 (9) Å] implies double bond character, while C4—O9 [1.332 (9) Å] is well regarded as a single bond. The dihedral angle between the two benzene ring mean planes (C1—C6 and C8—C13) is 32.2 (3)°. A network of hydrogen bonds helps to establish the crystal packing.

### Experimental

One mmol of zinc acetate was added to a solution of salicylaldehyde-4-aminobenzene sulfonic acid (1 mmol) in 20 ml of 95%  $CH_3CH_2OH$ . The mixture was continuously stirred for 2 h at refluxing temperature, evaporating some methanol, then, upon cooling, the solid product was collected by filtration and dried *in vacuo* (yield 76%). Clear blocks of (I) were obtained by evaporation from a methanol solution after a week.

### Refinement

The water H atoms were located in a difference map and refined as riding in their as-found relative positions with  $U_{iso}(H) = 1.2U_{eq}(O)$ . Other H atoms were placed geometrically (C—H = 0.93–0.97 Å, O—H = 0.82 Å, N—H = 0.86 Å) and refined as riding with  $U_{iso}(H) = 1.2U_{eq}(C,N)$  or  $1.5U_{eq}(O)$ .

### Figures



Fig. 1. The complex molecule in (I) with 50% probability ellipsoids (arbitrary spheres for the H atoms).

## Hexaaquazinc(II) bis[4-(2-hydroxybenzylideneamino)benzenesulfonate]

### Crystal data

$[Zn(H_2O)_6](C_{13}H_{10}NO_4S)_2$

$M_r = 726.03$

Monoclinic,  $P2_1/n$

$a = 6.3255$  (10) Å

$b = 35.312$  (3) Å

$c = 6.9832$  (10) Å

$F_{000} = 752$

$D_x = 1.546$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 2287 reflections

$\theta = 2.3$ – $23.1^\circ$

$\mu = 0.99$  mm<sup>-1</sup>

# supplementary materials

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$\beta = 90.391 (2)^\circ$   
 $V = 1559.8 (4) \text{ \AA}^3$   
 $Z = 2$

$T = 298 (2) \text{ K}$   
Block, colourless  
 $0.35 \times 0.33 \times 0.20 \text{ mm}$

## Data collection

Bruker SMART CCD area-detector diffractometer  
Radiation source: fine-focus sealed tube  
Monochromator: graphite  
 $T = 298(2) \text{ K}$   
phi and  $\omega$  scans  
Absorption correction: multi-scan (SADABS; Bruker, 1997)  
 $T_{\min} = 0.723$ ,  $T_{\max} = 0.826$   
6994 measured reflections

2708 independent reflections  
2170 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.034$   
 $\theta_{\text{max}} = 25.0^\circ$   
 $\theta_{\text{min}} = 2.3^\circ$   
 $h = -7 \rightarrow 5$   
 $k = -42 \rightarrow 32$   
 $l = -8 \rightarrow 8$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.069$   
 $wR(F^2) = 0.168$   
 $S = 1.08$   
2708 reflections  
205 parameters  
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map  
Hydrogen site location: inferred from neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.063P)^2 + 5.0428P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.37 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.79 \text{ e \AA}^{-3}$   
Extinction correction: none

## Special details

**Experimental.** 'SADABS v2.0 (Bruker, 1997)'

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{Å}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
Zn1	1.0000	0.0000	0.0000	0.0366 (3)

N1	0.4209 (9)	0.23101 (16)	0.5074 (8)	0.0590 (15)
O1	0.8397 (6)	0.06182 (11)	0.5063 (6)	0.0432 (10)
O2	0.5147 (6)	0.05114 (12)	0.6744 (6)	0.0517 (11)
O3	0.5147 (6)	0.05133 (12)	0.3282 (6)	0.0508 (11)
O4	0.1459 (9)	0.28426 (15)	0.5713 (9)	0.0841 (17)
H4	0.1765	0.2635	0.5271	0.126*
O5	0.7091 (6)	0.02586 (12)	0.0024 (6)	0.0508 (11)
H5A	0.6633	0.0365	-0.0989	0.061*
H5B	0.6403	0.0352	0.0953	0.061*
O6	1.1063 (6)	0.03918 (14)	-0.1984 (7)	0.0634 (13)
H6A	1.2325	0.0438	-0.2321	0.076*
H6B	1.0273	0.0452	-0.2926	0.076*
O7	1.1046 (6)	0.03394 (15)	0.2234 (6)	0.0659 (14)
H7A	1.0205	0.0428	0.3067	0.079*
H7B	1.2285	0.0364	0.2696	0.079*
S1	0.6114 (2)	0.06635 (4)	0.5026 (2)	0.0358 (4)
C1	0.5612 (8)	0.11535 (16)	0.4999 (7)	0.0352 (12)
C2	0.7196 (10)	0.14097 (18)	0.5569 (9)	0.0504 (16)
H2	0.8529	0.1323	0.5932	0.060*
C3	0.6734 (12)	0.17949 (19)	0.5582 (10)	0.0616 (19)
H3	0.7770	0.1968	0.5946	0.074*
C4	0.4756 (11)	0.19220 (18)	0.5058 (10)	0.0534 (16)
C5	0.3215 (11)	0.16672 (18)	0.4537 (10)	0.0577 (18)
H5	0.1871	0.1755	0.4217	0.069*
C6	0.3630 (9)	0.12829 (17)	0.4482 (9)	0.0486 (15)
H6	0.2584	0.1113	0.4100	0.058*
C7	0.5591 (11)	0.25692 (19)	0.4805 (9)	0.0551 (17)
H7	0.6977	0.2500	0.4540	0.066*
C8	0.5058 (12)	0.29691 (18)	0.4902 (10)	0.0567 (17)
C9	0.3022 (13)	0.3086 (2)	0.5377 (11)	0.067 (2)
C10	0.2618 (15)	0.3473 (2)	0.5562 (11)	0.072 (2)
H10	0.1281	0.3554	0.5926	0.086*
C11	0.4158 (15)	0.3734 (2)	0.5214 (11)	0.073 (2)
H11	0.3849	0.3991	0.5311	0.087*
C12	0.6194 (16)	0.3620 (2)	0.4714 (12)	0.082 (2)
H12	0.7243	0.3798	0.4477	0.098*
C13	0.6621 (13)	0.3239 (2)	0.4578 (11)	0.067 (2)
H13	0.7978	0.3160	0.4264	0.081*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Zn1	0.0234 (4)	0.0495 (6)	0.0370 (5)	-0.0006 (4)	-0.0002 (3)	-0.0018 (4)
N1	0.064 (4)	0.047 (3)	0.066 (4)	0.013 (3)	-0.002 (3)	-0.007 (3)
O1	0.0223 (18)	0.062 (3)	0.046 (2)	0.0040 (17)	0.0003 (16)	0.003 (2)
O2	0.037 (2)	0.066 (3)	0.052 (3)	0.003 (2)	0.0060 (19)	0.018 (2)
O3	0.032 (2)	0.064 (3)	0.057 (3)	0.0026 (19)	-0.0060 (19)	-0.011 (2)
O4	0.085 (4)	0.068 (3)	0.100 (5)	0.013 (3)	0.027 (3)	0.005 (3)

## supplementary materials

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O5	0.035 (2)	0.079 (3)	0.038 (2)	0.016 (2)	0.0023 (18)	0.002 (2)
O6	0.031 (2)	0.099 (4)	0.060 (3)	-0.005 (2)	-0.003 (2)	0.030 (3)
O7	0.029 (2)	0.109 (4)	0.060 (3)	-0.002 (2)	-0.004 (2)	-0.038 (3)
S1	0.0239 (7)	0.0472 (8)	0.0362 (7)	0.0032 (6)	0.0008 (5)	-0.0001 (6)
C1	0.028 (3)	0.047 (3)	0.031 (3)	0.005 (2)	-0.001 (2)	-0.005 (2)
C2	0.039 (3)	0.060 (4)	0.053 (4)	-0.005 (3)	-0.010 (3)	0.000 (3)
C3	0.066 (5)	0.049 (4)	0.069 (5)	-0.007 (3)	-0.007 (4)	-0.012 (3)
C4	0.053 (4)	0.051 (4)	0.056 (4)	0.012 (3)	0.010 (3)	-0.002 (3)
C5	0.046 (4)	0.048 (4)	0.079 (5)	0.007 (3)	-0.007 (3)	0.000 (3)
C6	0.037 (3)	0.047 (4)	0.062 (4)	0.005 (3)	-0.014 (3)	0.001 (3)
C7	0.058 (4)	0.055 (4)	0.053 (4)	0.008 (3)	0.001 (3)	-0.005 (3)
C8	0.068 (4)	0.048 (4)	0.053 (4)	0.003 (4)	-0.002 (3)	-0.014 (3)
C9	0.081 (6)	0.061 (5)	0.058 (5)	0.006 (4)	0.006 (4)	-0.001 (4)
C10	0.104 (7)	0.052 (4)	0.059 (5)	0.019 (4)	0.005 (4)	-0.003 (3)
C11	0.108 (7)	0.047 (4)	0.062 (5)	0.019 (4)	-0.006 (4)	-0.002 (4)
C12	0.109 (8)	0.052 (5)	0.084 (6)	0.002 (5)	-0.007 (5)	0.003 (4)
C13	0.073 (5)	0.059 (5)	0.071 (5)	0.002 (4)	-0.008 (4)	-0.007 (4)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

Zn1—O5 <sup>i</sup>	2.054 (4)	C1—C2	1.406 (8)
Zn1—O5	2.054 (4)	C2—C3	1.391 (9)
Zn1—O7 <sup>i</sup>	2.072 (4)	C2—H2	0.9300
Zn1—O7	2.072 (4)	C3—C4	1.376 (9)
Zn1—O6	2.073 (4)	C3—H3	0.9300
Zn1—O6 <sup>i</sup>	2.073 (4)	C4—C5	1.374 (9)
N1—C7	1.280 (9)	C5—C6	1.383 (9)
N1—C4	1.413 (8)	C5—H5	0.9300
O1—S1	1.453 (4)	C6—H6	0.9300
O2—S1	1.453 (4)	C7—C8	1.454 (9)
O3—S1	1.459 (4)	C7—H7	0.9300
O4—C9	1.332 (9)	C8—C13	1.394 (10)
O4—H4	0.8200	C8—C9	1.394 (10)
O5—H5A	0.8500	C9—C10	1.397 (10)
O5—H5B	0.8499	C10—C11	1.365 (11)
O6—H6A	0.8500	C10—H10	0.9300
O6—H6B	0.8500	C11—C12	1.396 (12)
O7—H7A	0.8500	C11—H11	0.9300
O7—H7B	0.8500	C12—C13	1.375 (10)
S1—C1	1.759 (6)	C12—H12	0.9300
C1—C6	1.380 (8)	C13—H13	0.9300
O5 <sup>i</sup> —Zn1—O5	180.0 (2)	C3—C2—H2	120.6
O5 <sup>i</sup> —Zn1—O7 <sup>i</sup>	91.04 (17)	C1—C2—H2	120.6
O5—Zn1—O7 <sup>i</sup>	88.96 (17)	C4—C3—C2	120.5 (6)
O5 <sup>i</sup> —Zn1—O7	88.96 (17)	C4—C3—H3	119.7
O5—Zn1—O7	91.04 (17)	C2—C3—H3	119.7
O7 <sup>i</sup> —Zn1—O7	180.0 (3)	C5—C4—C3	119.9 (6)

O5 <sup>i</sup> —Zn1—O6	89.80 (17)	C5—C4—N1	117.6 (6)
O5—Zn1—O6	90.20 (17)	C3—C4—N1	122.4 (6)
O7 <sup>i</sup> —Zn1—O6	89.2 (2)	C4—C5—C6	121.0 (6)
O7—Zn1—O6	90.8 (2)	C4—C5—H5	119.5
O5 <sup>i</sup> —Zn1—O6 <sup>i</sup>	90.20 (17)	C6—C5—H5	119.5
O5—Zn1—O6 <sup>i</sup>	89.80 (17)	C1—C6—C5	119.4 (6)
O7 <sup>i</sup> —Zn1—O6 <sup>i</sup>	90.8 (2)	C1—C6—H6	120.3
O7—Zn1—O6 <sup>i</sup>	89.2 (2)	C5—C6—H6	120.3
O6—Zn1—O6 <sup>i</sup>	180.0 (3)	N1—C7—C8	121.9 (7)
C7—N1—C4	121.6 (6)	N1—C7—H7	119.1
C9—O4—H4	109.5	C8—C7—H7	119.1
Zn1—O5—H5A	119.3	C13—C8—C9	119.6 (7)
Zn1—O5—H5B	129.9	C13—C8—C7	119.5 (7)
H5A—O5—H5B	106.9	C9—C8—C7	120.9 (7)
Zn1—O6—H6A	128.5	O4—C9—C8	122.6 (7)
Zn1—O6—H6B	119.5	O4—C9—C10	118.6 (8)
H6A—O6—H6B	106.6	C8—C9—C10	118.8 (8)
Zn1—O7—H7A	122.0	C11—C10—C9	120.9 (8)
Zn1—O7—H7B	129.4	C11—C10—H10	119.5
H7A—O7—H7B	106.4	C9—C10—H10	119.5
O1—S1—O2	111.6 (2)	C10—C11—C12	120.6 (7)
O1—S1—O3	112.7 (2)	C10—C11—H11	119.7
O2—S1—O3	112.2 (3)	C12—C11—H11	119.7
O1—S1—C1	106.7 (2)	C13—C12—C11	118.8 (8)
O2—S1—C1	107.2 (3)	C13—C12—H12	120.6
O3—S1—C1	105.9 (2)	C11—C12—H12	120.6
C6—C1—C2	120.4 (5)	C12—C13—C8	121.2 (8)
C6—C1—S1	119.5 (4)	C12—C13—H13	119.4
C2—C1—S1	120.1 (4)	C8—C13—H13	119.4
C3—C2—C1	118.8 (6)		
O1—S1—C1—C6	163.8 (5)	S1—C1—C6—C5	177.4 (5)
O2—S1—C1—C6	-76.5 (5)	C4—C5—C6—C1	1.4 (10)
O3—S1—C1—C6	43.5 (5)	C4—N1—C7—C8	-177.3 (6)
O1—S1—C1—C2	-18.6 (5)	N1—C7—C8—C13	-179.4 (7)
O2—S1—C1—C2	101.1 (5)	N1—C7—C8—C9	2.8 (11)
O3—S1—C1—C2	-138.9 (5)	C13—C8—C9—O4	-179.7 (7)
C6—C1—C2—C3	-0.7 (9)	C7—C8—C9—O4	-1.9 (11)
S1—C1—C2—C3	-178.3 (5)	C13—C8—C9—C10	-1.5 (11)
C1—C2—C3—C4	0.5 (10)	C7—C8—C9—C10	176.3 (7)
C2—C3—C4—C5	0.7 (11)	O4—C9—C10—C11	-179.3 (7)
C2—C3—C4—N1	179.1 (6)	C8—C9—C10—C11	2.5 (12)
C7—N1—C4—C5	-152.7 (7)	C9—C10—C11—C12	-1.7 (12)
C7—N1—C4—C3	28.8 (10)	C10—C11—C12—C13	0.0 (12)
C3—C4—C5—C6	-1.7 (11)	C11—C12—C13—C8	0.9 (12)
N1—C4—C5—C6	179.8 (6)	C9—C8—C13—C12	-0.1 (11)
C2—C1—C6—C5	-0.2 (9)	C7—C8—C13—C12	-178.0 (7)

Symmetry codes: (i)  $-x+2, -y, -z$ .

## supplementary materials

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### Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O7—H7B···O3 <sup>ii</sup>	0.85	1.93	2.760 (6)	166
O7—H7A···O1	0.85	1.93	2.779 (5)	177
O6—H6B···O1 <sup>iii</sup>	0.85	1.92	2.773 (6)	176
O6—H6A···O2 <sup>iv</sup>	0.85	1.92	2.770 (6)	175
O5—H5B···O3	0.85	1.90	2.745 (6)	171
O5—H5A···O2 <sup>iii</sup>	0.85	1.91	2.742 (6)	167
O4—H4···N1	0.82	1.93	2.602 (7)	139

Symmetry codes: (ii)  $x+1, y, z$ ; (iii)  $x, y, z-1$ ; (iv)  $x+1, y, z-1$ .



Fig. 1

